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## The mechanisms of mass-independent fractionation of sulfur isotopes

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The discovery of mass-independent fractionation (MIF) of sulfur isotopes in Archean and Paleoproterozoic sedimentary rocks (Farquhar et al. 2000a) has profoundly influenced our understanding of the sulfur cycle and the rise of  $O_2$  on early Earth. Sulfur MIF is expected to arise from gas-phase atmospheric reactions involving  $SO_2$  and  $H_2S$ . However, a quantitative understanding of the mechanisms remains elusive. Here I will discuss the most likely MIF mechanisms for sulfur isotopes, and use existing laboratory data (from the Thiemens and Farquhar laboratories) to place constraints on mechanism.

First, I give a brief description of the most relevant laboratory data: 1) Photolysis of H<sub>2</sub>S yields elemental sulfur (S<sub>el</sub>) with small  $\Delta^{33}$ S, and is primarily mass-dependently fractionated (Farquhar et al. 2000b); 2) Spark discharge of SO<sub>2</sub> yields S<sub>el</sub> with small  $\Delta^{33}$ S, and is primarily mass-dependent (Wing et al. 2004); 3) SO<sub>2</sub> photolysis from 190-210 nm yields S<sub>el</sub> with  $\Delta^{33}$ S ~ + 20 %, along a  $\delta^{33}$ S/ $\delta^{34}$ S slope ~ 0.6, which is a large MIF signature (Wing et al. 2004); 4) SO<sub>2</sub> photolysis at wavelengths > 220 nm yields SO<sub>4</sub><sup>2-</sup> with  $\Delta^{33}$ S ~ 3 %, along a  $\delta^{33}$ S/ $\delta^{34}$ S slope ~ 0.6 (Farquhar et al. 2001), also a significant MIF signature.

I will interpret this data within the context of two MIF mechanisms: a) non-RRKM effects, as suggested for  $O_3$  (Gao and Marcus 2001), and b) isotope-selective photodissociation accompanying predissociation. Two additional mechanisms are hyperfine effects and near-resonant spin-orbit coupling (Bhattacharya et al. 2000), but here I will focus on a) and b). Non-RRKM effects result from a lack of intramolecular equilibrium within a vibrationally excited complex. In the case of  $O_3$ , Gao and Mar-

cus (2001) proposed that non-RRKM effects are strongest in symmetric isotopomers of  $O_3^*$ . The isoelectronic sulfur reaction,  $S + S_2 \rightarrow S_3$ , may also exhibit non-RRKM effects. However, the higher bond energy of  $S_3$  (2.7 eV) versus  $O_3$  (1.1 eV) may favor stabilization of  $S_3^*$ , and therefore diminish the magnitude of the MIF signature. Additionally, if the rate of the competing reaction,  $S_2 + S_2 \rightarrow S_4$ , is fast compared to  $S + S_2 \rightarrow S_3$  during formation of  $S_{el}$ , then any MIF during  $S_3$  formation may be of negligible importance. Also, if  $S_{el}$  forms primarily in surface reactions (e.g., walls) in the experiments, we can expect rapid stabilization of  $S_3^*$  and a diminished MIF effect. Thus, sulfur MIF during  $S_3$  formation is not assured.

Mass-independent isotope-selective photodissociation is likely in SO (~ 200 nm) and SH (>250 nm), possible in SO<sub>2</sub>, and unlikely in H<sub>2</sub>S. The lack of a MIF signature for S<sub>el</sub> produced by long-exposure H<sub>2</sub>S photolysis results, in part, from the mass-dependence of H<sub>2</sub>S + hv  $\rightarrow$  SH + H.

I will present kinetics calculations of the above  $H_2S$  and  $SO_2$  photolysis experiments, and show how it is possible to use the results of these experiments to constrain the mechanism of MIF for atmospheric sulfur species.